Carboxylate Linked Dimeric and Polymeric Coordination Compounds

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Introduction

 C_3 -symmetric triaminoguanidinium based ligands are able to coordinate hard metal ions like Ti(IV) or Zr(IV) as well as soft metal ions like Cd(II) or Pd(II) in their tris-chelating binding pockets. A variety of different discrete supramolecular coordination cages such as tetrahedra, octahedra and trigonal bipyramids are accessible.^[1] The modification of salicylic aldehyde into pyridine-N-oxide based ligands reduces the negative charge of the deprotonated species.^[2]

Coordination of Zn(II) in the presence of carboxylates and pseudohalides like azide or thiocyanate gives cyclic coordination oligomers, which can function as supramolecular host molecules for fullerene $C_{60/70}$ encapsulation.^[2,3] The absence of (pseudo)halides leads to the formation of a series of different dimeric and polymeric structures depending on the carboxylate chain length. They are further discussed on this poster.

Ligand Synthesis and Zn(II) Coordination



• its charge is compensated by further coligand coordination like carboxylates or (pseudo)halides

Zn(II)-Zn(II) Linkage by Carboxylates





References

[1]

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- C. R. Göb, *Dissertation*, RWTH Aachen, 2018. [2]
- C. R. Göb, L. Sturm, Y. Tobe, I. M. Oppel, manuscript submitted. [3]